

## Unraveling the Dynamic Structural Evolution of Phthalocyanine Catalysts during CO<sub>2</sub> Electroreduction

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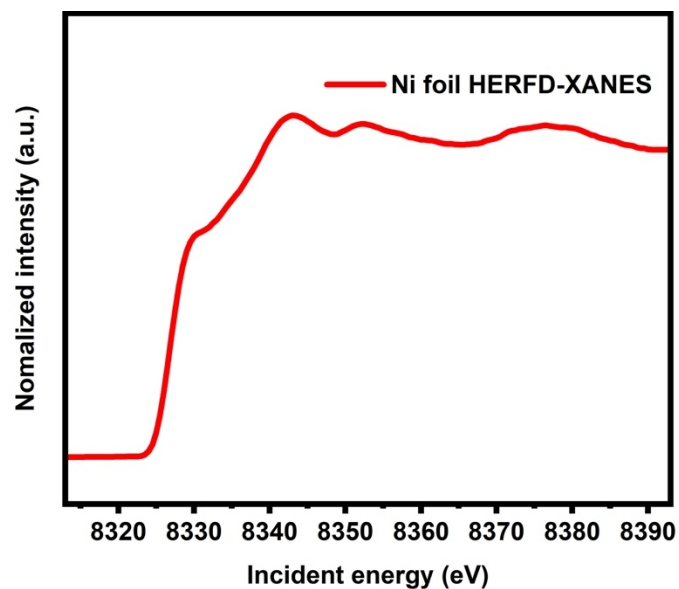
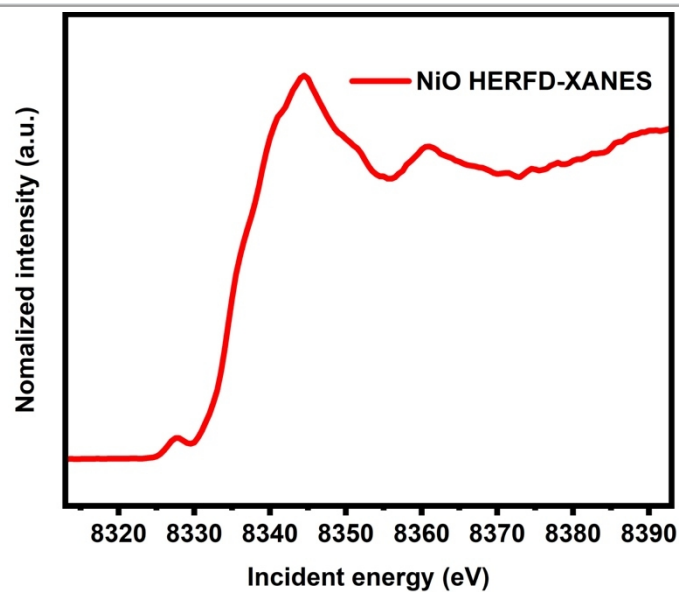
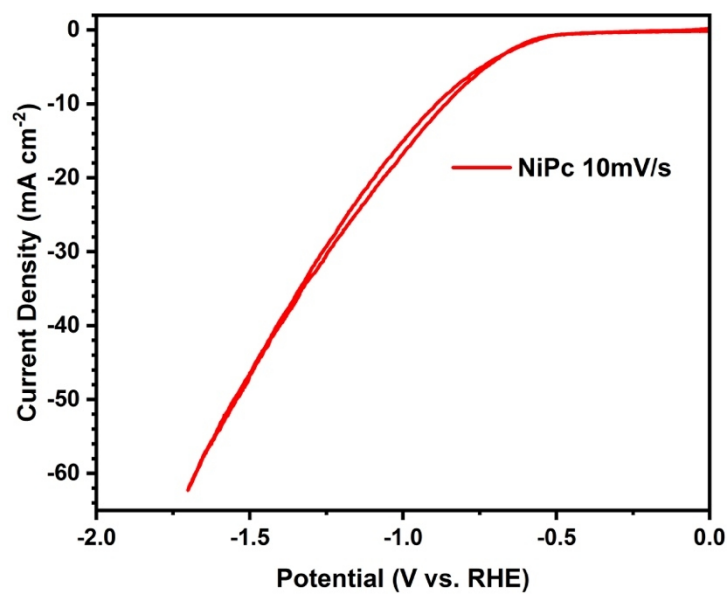


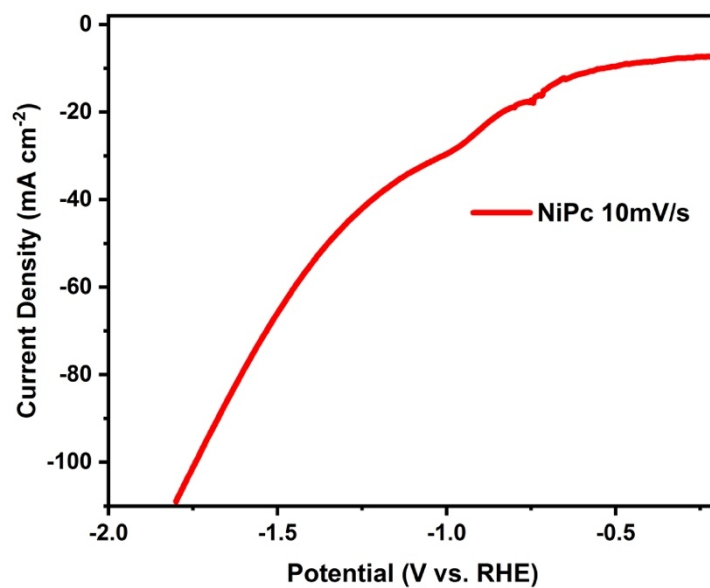
Figure S1. The Ni HERFD-XANES spectrum obtained by integrating the RIXS spectra.



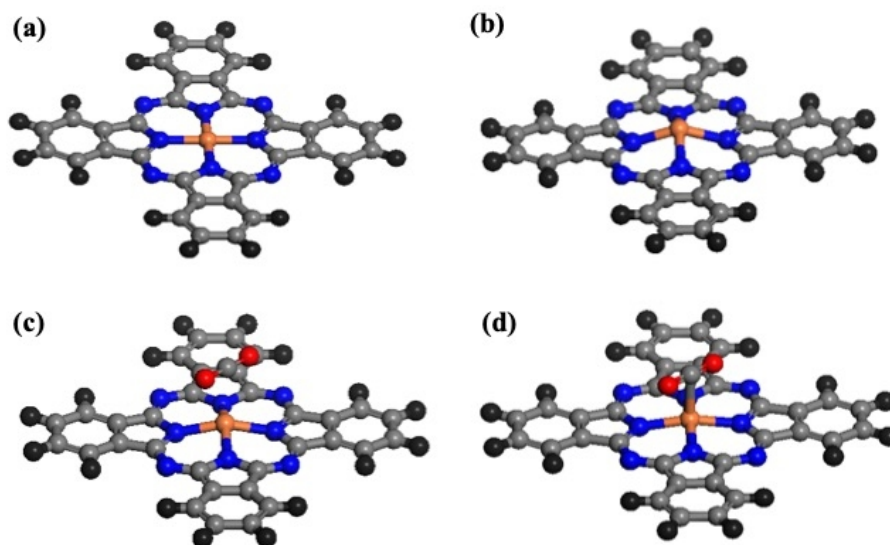
**Figure S2.** The NiO HERFD-XANES spectrum obtained by integrating the RIXS spectra.



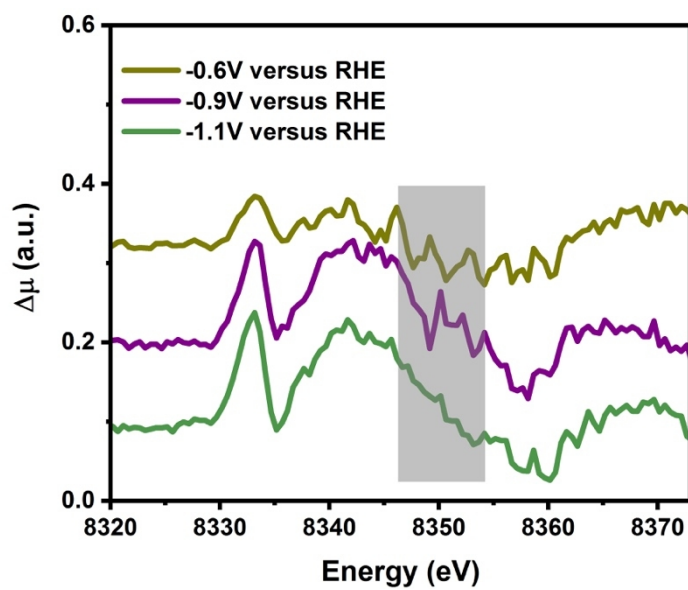
**Figure S3.** Cyclic voltammetry curves of NiPc at a scan rate of 10 mV s<sup>-1</sup> in CO<sub>2</sub> saturated 0.1 M KHCO<sub>3</sub> aqueous solution.



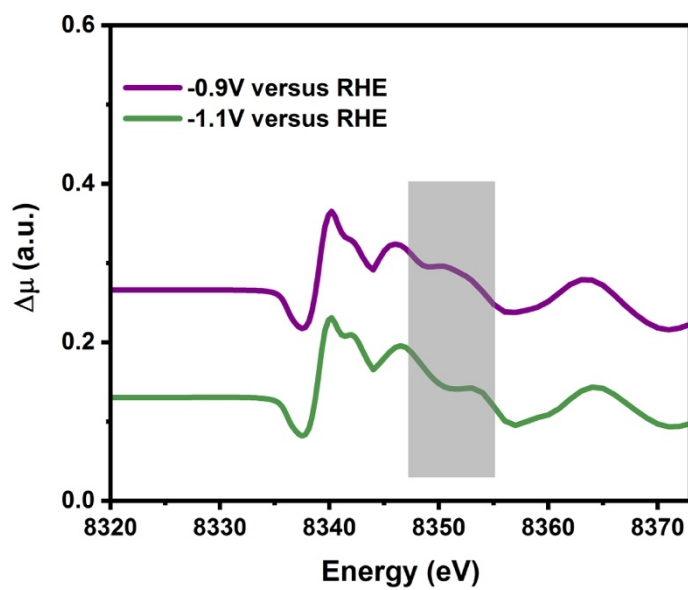
**Figure S4.** Linear sweep voltammetry curves of NiPc at a scan rate of  $10 \text{ mV s}^{-1}$  in  $\text{CO}_2$  saturated  $0.1 \text{ M KHCO}_3$  aqueous solution.



**Figure S5.** The schemes show the corresponding calculation model for the XANES simulation (Ni (orange), N (blue), C (gray), O (red) and H (black)).

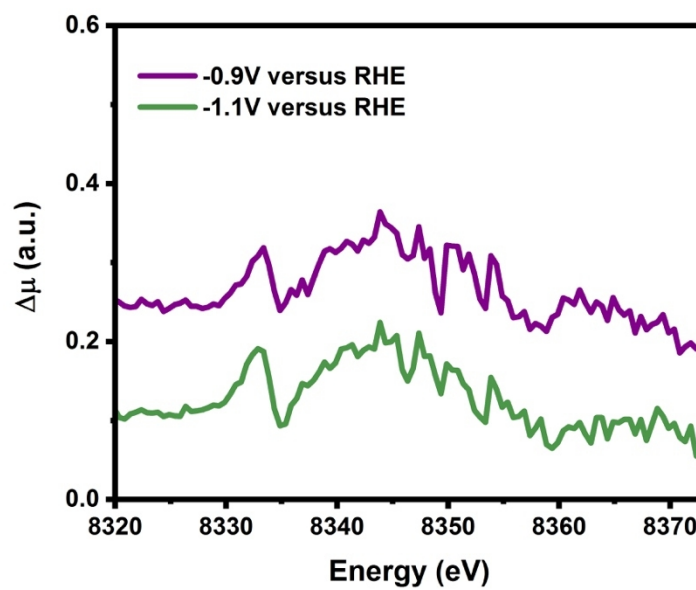


**Figure S6.** *Operando* Ni K-edge  $\Delta\mu$  spectra of NiPc at various potentials vs. RHE by subtracting the wet HERFD-XANES spectrum.

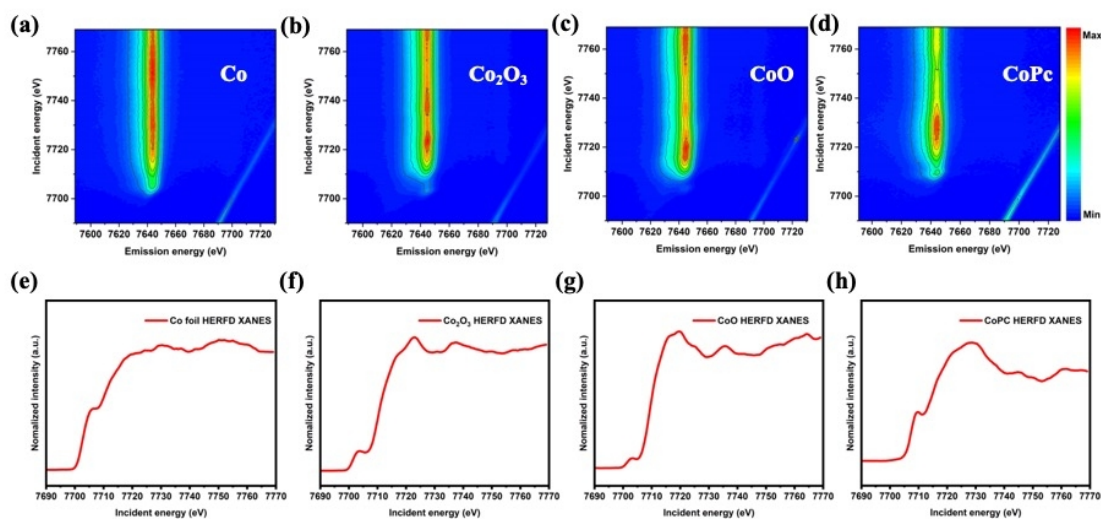


**Figure S7.** Theoretical Ni K-edge  $\Delta\mu$  spectra of NiPc at various potentials vs. RHE by subtracting the wet HERFD-XANES spectrum.

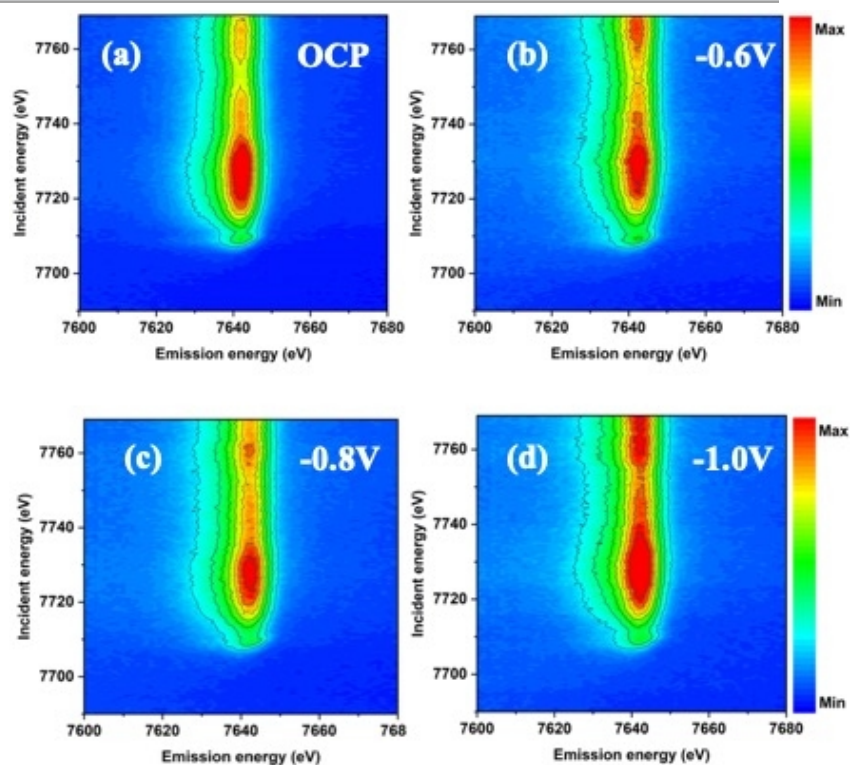




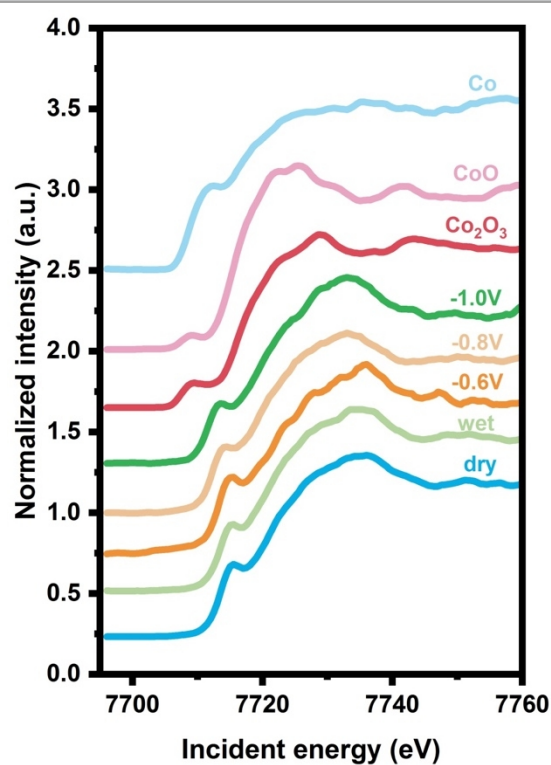
**Figure S8.** Operando Ni K-edge  $\Delta\mu$  spectra of NiPc at various potentials vs. RHE by subtracting the HERFD-XANES spectrum at the 0.6V versus RHE.



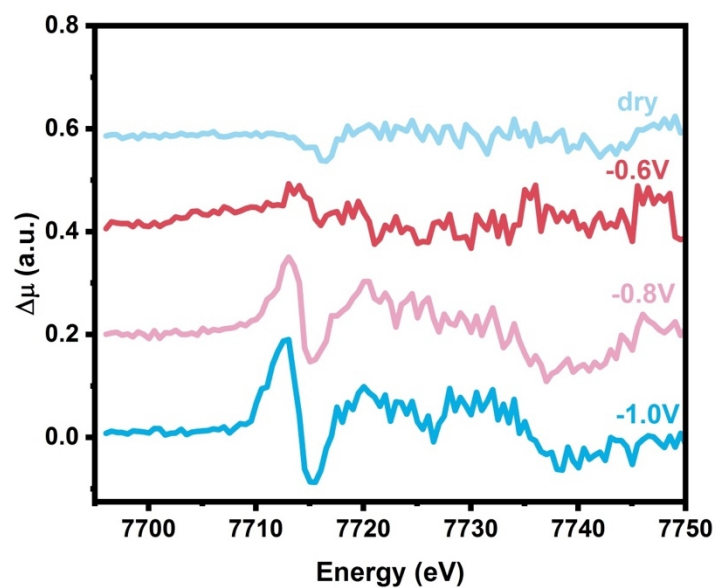
**Figure S9.** (a)  $K\beta_{1,3}$ -RIXS plane of Co foil. The contour plane at the emission energy of  $\sim 7644$  eV is the  $K\beta_{1,3}$ -RIXS plane. (b)  $K\beta_{1,3}$ -RIXS plane of Co<sub>2</sub>O<sub>3</sub>. (c)  $K\beta_{1,3}$ -RIXS plane of CoO. (d)  $K\beta_{1,3}$ -RIXS plane of CoO. (e-f) HERFD-XANES spectra of Co foil, Co<sub>2</sub>O<sub>3</sub>, CoO and CoPc. The HERFD-XANES spectrum was obtained by integrating the RIXS intensity.



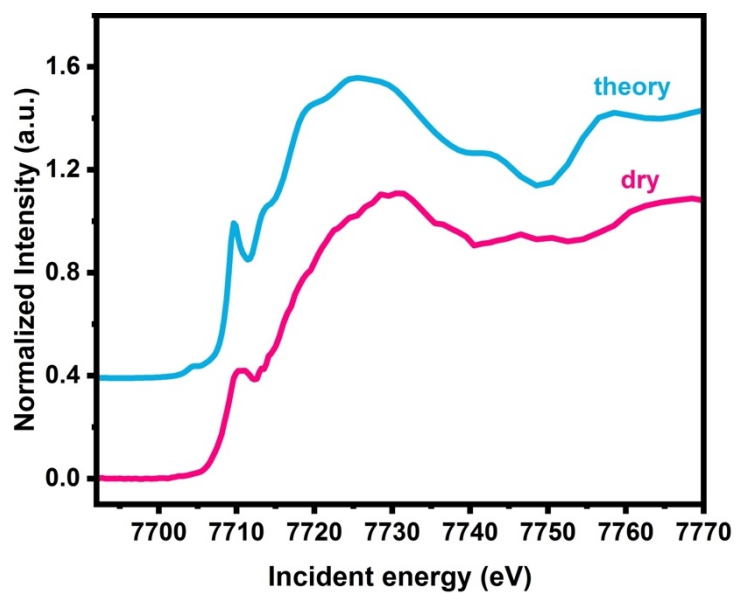
**Figure S10.** Operando  $K\beta_{1,3}$ -RIXS planes of CoPc without potential applied (a) and NiPc under (b-d) various potentials vs. RHE in a 0.5 M  $\text{KHCO}_3$  solution during the  $\text{ECO}_2\text{RR}$ .



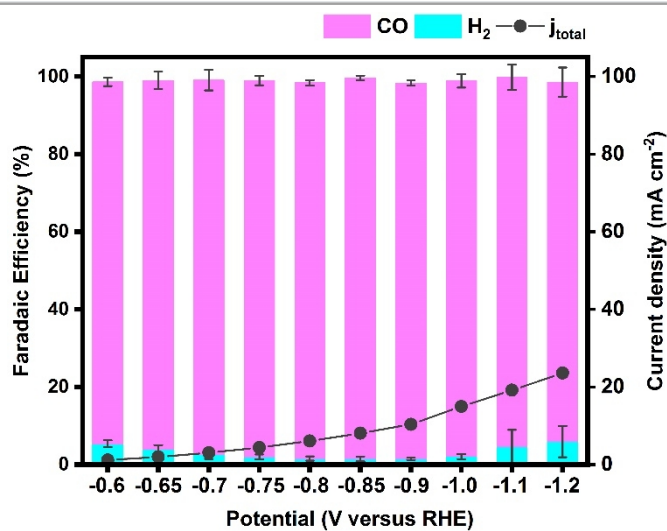
**Figure S11.** Normalized K-edge HERFD-XANES of CoPc at various applied potentials vs. RHE during the CO<sub>2</sub>RR and corresponding standard references of Co, Co<sub>2</sub>O<sub>3</sub> and CoO.



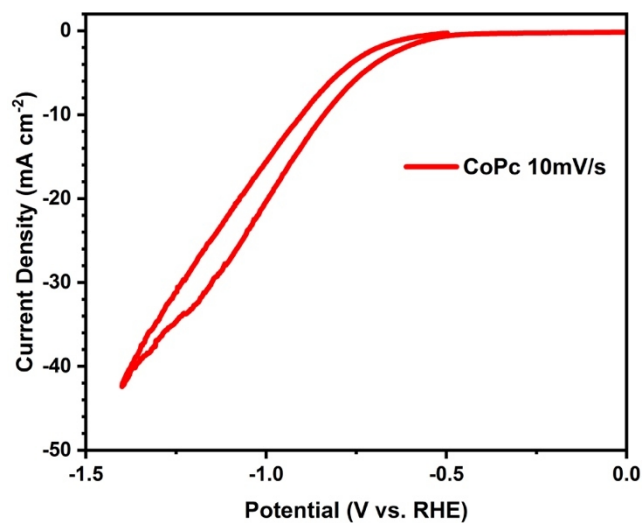
**Figure S12.** Operando Co K-edge  $\Delta\mu$  spectra of CoPc at various potentials vs. RHE by subtracting the dry HERFD-XANES spectrum.



**Figure S13.** Theoretical HERFD-XANES spectrum of CoPc (blue) and the normalized K-edge HERFD-XANES spectrum of CoPc (pink).

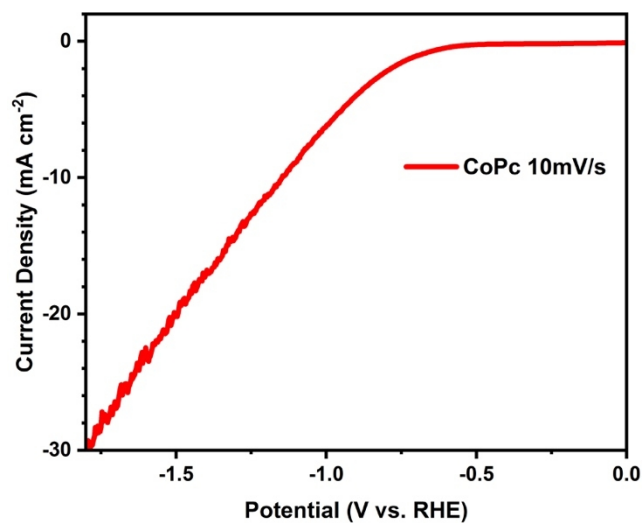


**Figure S14.** Faradaic efficiency and total current density of CoPc at the applied potentials (0.5 M KHCO<sub>3</sub> saturated by CO<sub>2</sub>). The faradaic efficiency is stated as an average and calculated at the steady-state current and product concentration.



**Figure S15.** Cyclic voltammetry curves of CoPc at a scan rate of  $10 \text{ mV s}^{-1}$  in  $\text{CO}_2$  saturated  $0.1 \text{ M KHCO}_3$  aqueous solution.





**Figure S16.** Linear sweep voltammetry curves of CoPc at a scan rate of 10 mV s<sup>-1</sup> in CO<sub>2</sub> saturated 0.1 M KHCO<sub>3</sub> aqueous solution.

**Table 1.** Electron Number and d-Band Center of NiPc, Distorted NiPc and NiPc-CO<sub>2</sub>

Model	Electron number	d-Band center
NiPc	6.39	-0.93
Distorted NiPc	7.28	-1.25
NiPc-CO <sub>2</sub>	2.71	-1.05